

Thermodynamic Study of Carbon Mineralization with Recycled Concrete Fines for Carbon Capture and Utilization Applications

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I. Introduction

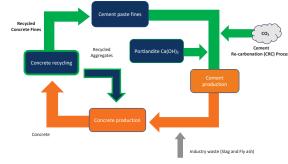


Figure 1.1: Concrete life cycle with lower CO $_2$ emissions. The green coloring represents the cycle improvements on the original orange process.

Motivation

- Over 30 billion tonnes of concrete are produced each year contributing to almost 10% of global CO₂ emissions.
- Although recycled concrete aggregates are more frequently used in practice, recycled concrete fines are mostly wasted.
- Carbon Mineralization is the process that occurs when concrete is exposed to carbon dioxide in the atmosphere or synthetically.
- Using the concrete as a form of Carbon Capture and Utilization (CCUS) will reduce the net CO₂ emissions produced.



Figure 1.2 : Comparison of current concrete production process (1) with novel concrete process (2)

- Calcination of Limestone is the largest contributor to CO₂ emissions.
- limestone co co
- The addition of cementitious supplementary materials (SCM) and CO_2 as a part of the cement mixture provides a pathway to use industry waste.
- Portlandite is an alkali activator that improves the reaction environment.

II. Materials and Methods

Modified Parrot-Killoh (MPK)

 Takes the non equilibrium inputs of the clinker phases and a few oxides and computes the mass of each at a given time based on their dissolution rates

1 + (x/c)

Figure 2.1: 5PL Equation

5 Parameter Logistic (5PL)

- The 5PL fit is used to model the dissolution of the recycle concrete fines based on the degree of reaction (DoR).
- 5PL parameters are traditionally calculated based on time-variant QXRD data.

CemGEMS

• Thermodynamic modeling using the Parrot-Killoh method for plots describing the evolution of hydration product volumes with respect to time. [1]

Recycled Concrete Fines

- The composition of recycled concrete is dependent on the location and construction practices of that region.
- For this study, the average of three Dutch concrete plants were used:

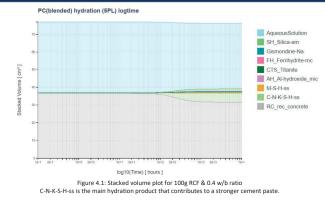


III. Discussion

RCF 4PL/5PL Parameters

- Due to a lack of QXRD data for RCF to fit the parameters, we used heat of reaction data over a span of 24 hours to contrive the reaction extent trend.
- Finding a range of plausible values based off of the rules that govern the 4PL/5PL fit provided us with a reasonable set of parameters to model the hydration of RCF.
- Using CemGEMS, we are able to visually represent the reactants consumed and the products formed as a function of time.

IV. Preliminary Results



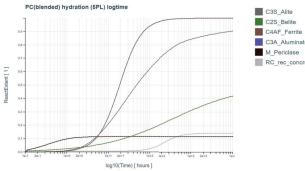


Figure 4.2: Plot containing the hydration reaction extent of clinker phases (alite, belite, ferrite, aluminate) and the extent of the recycled concrete fines. Up to 15% of the RCF is able to react to form hydration products.

V. Conclusion/ Future Work

- Substituting cement for RCF produces the main hydration products necessary.
- How much CO2 and portlandite at what temperature allows for the optimal process kinetics and products?
- What SCM will benefit the hydration process and concrete properties?
- How well do the experimental results match with the simulated data?

VI. Acknowledgements and References

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